

# Stepwise Synthesis of Fullerene Cyclopentadienide $R_5C_{60}^-$ and Indenide $R_3C_{60}^-$ . An Approach to Fully Unsymmetrically Substituted Derivatives

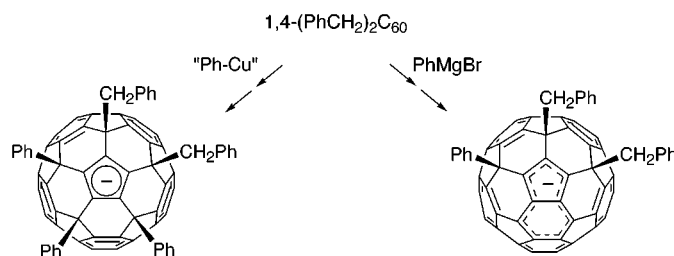
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Received April 27, 2000

## ABSTRACT



Fullerene cyclopentadienide  $(PhCH_2)_2PhC_{60}^-$  and indenide  $(PhCH_2)_2PhC_{60}^-$ , each bearing two different organic groups, were efficiently synthesized through regioselective reactions of 1,4- $(PhCH_2)_2C_{60}$  with an organocopper reagent ( $PhMgBr/CuBr \cdot SME_2$ ) or a Grignard reagent ( $PhMgBr$ ) followed by deprotonation with  $KO^tBu$ .

We reported earlier the synthesis of cyclopentadienes ( $R_5HC_{60}$ ;  $R = Ar, Me$ ) and cyclopentadienides ( $R_5C_{60}^-$ ) embedded in  $C_{60}$  through quantitative 5-fold organocopper addition to  $C_{60}$ .<sup>1</sup> The pentakis addition is so effective that a variety of  $R_5C_{60}^-$  compounds have been prepared, in many cases, in a quantitative yield; nonetheless, the method is totally unsuitable for the synthesis of compounds bearing different  $R$  groups or for the synthesis of  $C_{60}$  derivatives bearing fewer than five substituents. Here we report strategies for the installation of two different  $R$  groups totaling three and five in number as demonstrated through synthesis of new cyclopentadienide and indenides,  $K^+[(PhCH_2)_2Ph_3C_{60}]^-$  (**3**) and  $K^+[(PhCH_2)_2PhC_{60}]^-$  (**5**).

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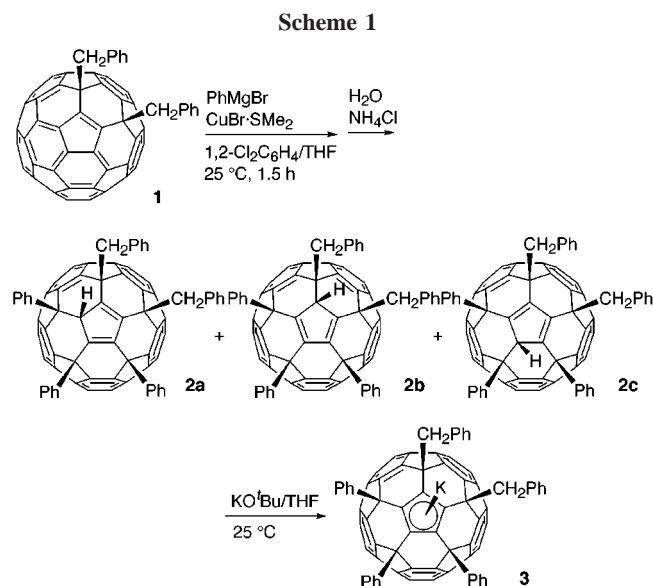
(1) (a) Sawamura, M.; Iikura, H.; Nakamura, E. *J. Am. Chem. Soc.* **1996**, *118*, 12850–12851. (b) Sawamura, M.; Iikura, H.; Ohama, T.; Hackler, U. E.; Nakamura, E. *J. Organomet. Chem.* **2000**, *599*, 32–36. (c) Sawamura, M.; Toganoh, M.; Kuninobu, Y.; Kato, S.; Nakamura, E. *Chem. Lett.* **2000**, 262–263. (d) Sawamura, M.; Iikura, H.; Hirai, A.; Nakamura, E. *J. Am. Chem. Soc.* **1998**, *120*, 8285–8286.

We are currently postulating that the 5-fold organocopper addition to  $C_{60}$  proceeds through a sequential addition/oxidation process involving nucleophilic attacks of  $R_2Cu^-$  species and one-electron oxidation by a  $Cu(I)$  salt. Formation of a 1,4-diorgano-1,4-dihydro[60]fullerene ( $1,4-R_2C_{60}$ )<sup>2</sup> and 1,4,11,30-tetraorgano-1,4,11,30-tetrahydro-[60]fullerene ( $1,4,11,30-R_4C_{60}$ )<sup>3</sup> as transient intermediates are being postulated. Along such a line of reasoning, we considered that the reaction of 1,4- $R_2C_{60}$  with an organocopper reagent

(2) (a) Nagashima, H.; Terasaki, H.; Kimura, E.; Nakajima, K.; Itoh, K. *J. Org. Chem.* **1994**, *59*, 1246–1248. (b) Subramanian, R.; Kadish, M. K.; Vijayashree, N. M.; Gao, X.; Jones, T. M.; Miller, D. M.; Krause, K. L.; Suenobu, T.; Fukuzumi, S. *J. Phys. Chem.* **1996**, *100*, 16327–16335. (c) Miki, S.; Kitao, M.; Fukunishi, K. *Tetrahedron Lett.* **1996**, *37*, 2049–2052. (d) Kitagawa, T.; Tanaka, T.; Tanaka, Y.; Takeuchi, K.; Komatsu, K. *J. Org. Chem.* **1995**, *60*, 1490–1491. (e) Wang, G.-W.; Murata, Y.; Komatsu, K.; Wan, T. S. M. *J. Chem. Soc., Chem. Commun.* **1996**, 2059–2060. (f) Fukuzumi, S.; Suenobu, T.; Hirasaka, T.; Arakawa, R.; Kadish, M. M. *J. Am. Chem. Soc.* **1998**, *120*, 9220–9227.

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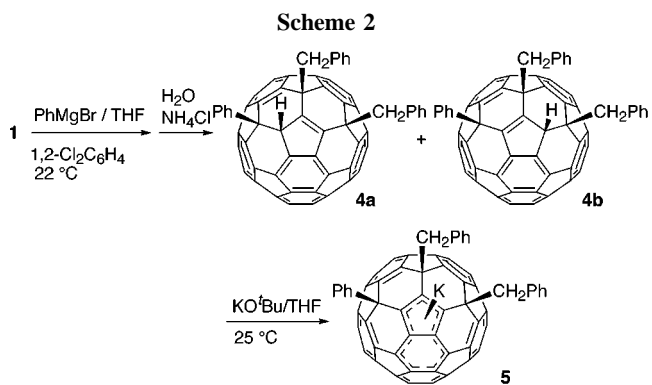
$R'_2Cu^-$  would afford the mixed organohydrofullerene  $R_2R'_3HC_{60}$ . We have chosen  $(PhCH_2)_2C_{60}$  (**1**)<sup>2b,c,f</sup> as a starting material of the present studies because of its availability and the high solubility in organic solvent. To our satisfaction, the reaction of **1** (1,2-dichlorobenzene/THF, 25 °C, 2 h) with an excess amount of an organocopper reagent prepared from PhMgBr (15 equiv) and CuBr·SMe<sub>2</sub> (15 equiv) (after quenching with aqueous NH<sub>4</sub>Cl) gave  $(PhCH_2)_2Ph_3H$  (**2a–c**) quantitatively as a mixture of three isomers (Scheme 1).



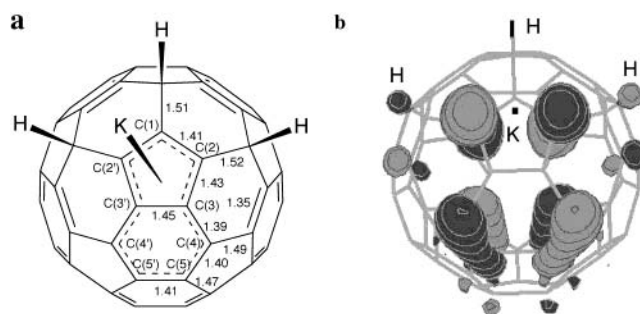
Purification by HPLC (Bucky Prep, Nacalai Tesque Co.) afforded an analytically pure material in 86% yield. Treatment of the solution of the isomeric mixture of **2** in THF with an excess KOtBu (1.2 equiv) at 20 °C caused immediate color change from dark red to very dark brown. The <sup>1</sup>H and <sup>13</sup>C NMR (THF-*d*<sub>8</sub>) showed that the isomeric mixture converged into a single cyclopentadienide  $K^+[(PhCH_2)_2Ph_3C_{60}]^-$  (**3**) with *C*<sub>s</sub> symmetry. This result indicates that the isomerism in **2** arises from the positional isomers of the cyclopentadiene moiety as shown for **2a–c** and, therefore, that the three phenyl groups were regioselectively installed in such a manner that one of the C<sub>60</sub> pentagons became isolated from the remainder of the C<sub>60</sub> core to form a cyclopentadiene.

It is known that a Grignard reagent reacts with C<sub>60</sub> in a 1,2-monoaddition manner.<sup>4</sup> We have conjectured that the reaction of 1,4-R<sub>2</sub>C<sub>60</sub> with a Grignard reagent will take place only once to give a triorganomonohydrofullerene. In light of our previous knowledge on a regioselectivity in the 3-fold addition reaction of C<sub>70</sub>, we considered that the reaction would give **5**. According to this scenario, we have carried out the reaction of 1,4- $(PhCH_2)_2C_{60}$  (**1**) with PhMgBr. The reaction of **1** proceeded smoothly in 1,2-dichlorobenzene at 22 °C in the presence of 3.6 equiv of PhMgBr (1 M in THF) and, after hydrolysis with aqueous NH<sub>4</sub>Cl and HPLC

purification (Bucky Prep), gave  $(PhCH_2)_2PhHC_{60}$  (**4a,b**) in 72% yield as a mixture of two isomers (100% conversion of C<sub>60</sub>). The structure of  $(PhCH_2)_2PhHC_{60}$  was assigned in the manner similar to the one employed for **2**; namely, deprotonation with KOtBu formed a single *C*<sub>1</sub> symmetric potassium salt  $K^+[(PhCH_2)_2PhC_{60}]^-$  (**5**) (Scheme 2).



Optimized structure of a *C*<sub>s</sub> symmetric model compound C<sub>60</sub>H<sub>3</sub>K (**6**) (HF/3-21G<sup>(\*)</sup>, total energy = −2857.00417344 hartree, Figure 1a)<sup>5</sup> shares some important features with that



**Figure 1.** (a) Part of the optimized structure of **6** at the HF/3-21G<sup>(\*)</sup> level. C–C bond lengths (Å) are affixed to the respective bonds. K–C bond lengths (Å): K–C(1), 3.08; K–C(2), 2.98; K–C(3), 2.93. (b) HOMO surface of **6**.

of its C<sub>70</sub> analogue, C<sub>70</sub>H<sub>3</sub>Li reported previously.<sup>1d</sup> Thus, the fully delocalized 10- $\pi$  indenyl framework is separated from the remaining 48- $\pi$  electron system by the seven surrounding C–C bonds of rather long bond lengths (1.47–1.52 Å).<sup>6</sup> The pentagon of the indenyl moiety coordinates to the metal center in an  $\eta^5$ -fashion. The HOMO surface of **6** depicted in Figure 1b represents a typical indenyl HOMO and shows remarkable resemblance to that of C<sub>70</sub>H<sub>3</sub>Li. Such an effective  $\pi$  electron delocalization within the indenyl framework of **6**

(5) Hehre, W.; Radom, L.; Schleyer, P. v. R.; Pople, J. A. *Ab Initio Molecular Orbital Theory*; Wiley: New York, 1986.

(6) The most obvious difference is found in the shorter C(3)–C(4) bond length (1.39 Å) of **6**.

(7) Note that the indenyl plane of C<sub>70</sub>H<sub>3</sub>Li is rather flat due to the decreased degree of curvature in the equatorial belt region of C<sub>70</sub> cage.

(4) (a) Fagan, P. J.; Krusic, P. J.; Evans, D. H.; Lerke, S. A.; Johnston, E. *J. Am. Chem. Soc.* **1992**, *114*, 9697–9699. (b) Hirsch, A.; Grösser, T.; Skiebe, A.; Soi, A. *Chem. Ber.* **1993**, *126*, 1061–1067.

is surprising if one consider the highly bent structure of the indenyl plane.<sup>7</sup>

In conclusion, we have shown that the cyclopentadiene ( $R^1R^2R^3HC_{60}$ ,  $R^1R^2HC_{60}$ ) and cyclopentadienide ( $R^1R^2R^3C_{60}^-$ ,  $R^1R^2C_{60}^-$ ) bearing two different organic substituents can efficiently be synthesized by stepwise methods by way of 1,4-diorganofullerene.<sup>2</sup> Since unsymmetrical 1,4-diorganofullerene  $RR'C_{60}$  of various combination of R and R' can be prepared through trapping of  $RC_{60}^-$  with an electrophile,<sup>2d-f</sup> we can fully expect that one can now synthesize a wide variety of analogues of  $R_5HC_{60}$  and  $R_3HC_{60}$  bearing various different organic substituents. The structure diversity thus created on a spherical platform will open new opportunities

on application of the densely functionalized, geometrically well-defined compounds in various fields.

**Acknowledgment.** This work was supported by Monbusho, Grant-in-Aid for Scientific Research on Priority Area (No. 283, Innovative Synthetic Reactions).

**Supporting Information Available:** Experimental procedures and characterization for compounds **2–5**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

OL005993K